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Key indicators

Single-crystal X-ray study
T = 193 K
Mean $\sigma(\text{C—C}) = 0.004 \text{ \AA}$
R factor = 0.030
wR factor = 0.073
Data-to-parameter ratio = 13.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

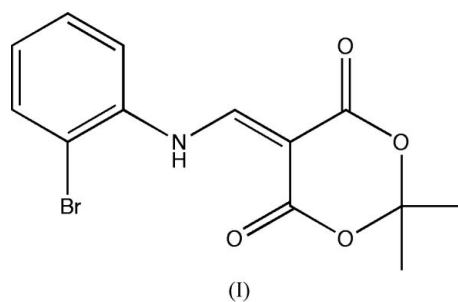
5-[(2-Bromophenylamino)methylene]-2,2-dimethyl-1,3-dioxane-4,6-dione

In the title compound, $\text{C}_{13}\text{H}_{12}\text{BrNO}_4$, the 1,3-dioxane-4,6-dione ring exhibits a half-chair conformation. The molecules are linked by $\text{C—H}\cdots\text{O}$ hydrogen bonds. The amino H atom has one intramolecular contact to a carbonyl O atom, forming a six-membered ring.

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Comment

Recently, we reported the structure of two arylaminomethylene Meldrum's acid derivatives (Joussef *et al.*, 2005*a,b*). As part of a continuing study of the conformational analysis in the solid state of Meldrum's acid derivatives, we report here the structure of the title compound, (I).



In (I), the 1,3-dioxane-4,6-dione ring exhibits a half-chair conformation. The dihedral angle C6—N7—C8—C9 is $170.5(3)^\circ$ and the distances C6—N7 and C8—C9 indicate delocalization. The amino H atom has one intramolecular contact to atom O10, with an $\text{H}\cdots\text{O}$ distance of $2.12(3) \text{ \AA}$, forming a six-membered ring. The delocalization of the N-atom lone pair into the Meldrum's acid ring may be favoured in the direction of one of the two carbonyl groups (Blake *et al.*, 2003). One intramolecular contact between the amino H atom and Br, with a distance of $2.62(3) \text{ \AA}$, is also observed. The molecular packing of (I) is stabilized by $\text{C—H}\cdots\text{O}$ hydrogen

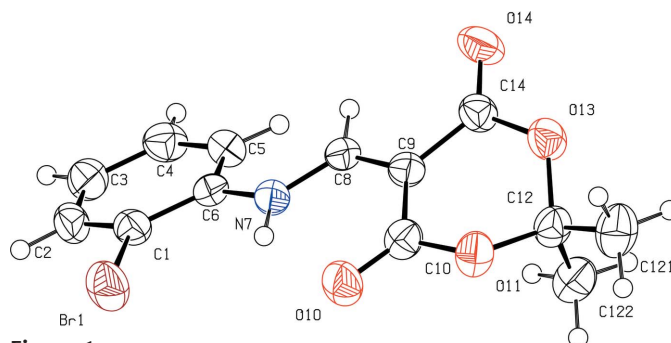


Figure 1

The molecular structure of (I), showing the atom labelling and displacement ellipsoids drawn at the 50% probability level.

bonds. The bond lengths and angles are in good agreement with values retrieved from the Cambridge Structural Database (Version 5.25; Allen, 2002). Details of the hydrogen bonding are given in Table 1.

Experimental

The title compound was prepared according to a literature procedure (Cassis *et al.*, 1985) and was recrystallized from methanol.

Crystal data

$C_{13}H_{12}BrNO_4$	$Z = 2$
$M_r = 326.15$	$D_x = 1.674 \text{ Mg m}^{-3}$
Triclinic, $P\bar{1}$	Mo $K\alpha$ radiation
$a = 6.960 (5) \text{ \AA}$	Cell parameters from 25 reflections
$b = 9.298 (5) \text{ \AA}$	$\theta = 4.1\text{--}16.9^\circ$
$c = 10.424 (5) \text{ \AA}$	$\mu = 3.19 \text{ mm}^{-1}$
$\alpha = 95.570 (5)^\circ$	$T = 193 (2) \text{ K}$
$\beta = 101.043 (5)^\circ$	Irregular block, colorless
$\gamma = 99.508 (5)^\circ$	$0.43 \times 0.26 \times 0.16 \text{ mm}$
$V = 647.2 (7) \text{ \AA}^3$	

Data collection

Enraf–Nonius CAD-4 diffractometer	$R_{\text{int}} = 0.020$
ω - 2θ scans	$\theta_{\text{max}} = 25.1^\circ$
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	$h = -8 \rightarrow 8$
$T_{\text{min}} = 0.417$, $T_{\text{max}} = 0.601$	$k = -11 \rightarrow 11$
2451 measured reflections	$l = -12 \rightarrow 0$
2312 independent reflections	3 standard reflections every 200 reflections
1724 reflections with $I > 2\sigma(I)$	intensity decay: 1%

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0298P)^2 + 0.241P]$
$R[F^2 > 2\sigma(F^2)] = 0.030$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.073$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.06$	$\Delta\rho_{\text{max}} = 0.32 \text{ e \AA}^{-3}$
2312 reflections	$\Delta\rho_{\text{min}} = -0.39 \text{ e \AA}^{-3}$
177 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.024 (2)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$N7\text{--}H7\cdots O10$	0.83 (3)	2.12 (3)	2.734 (3)	130 (3)
$C2\text{--}H2\cdots O14^i$	0.93	2.46	3.266 (4)	146

Symmetry code: (i) $x, y, z - 1$.

The amino H atom was located in a difference map and was refined freely. C-bound H atoms were positioned with idealized geometry and were refined as riding, with $C\text{--}H = 0.93 \text{ \AA}$ (0.96 \AA for methyl groups) and $U^{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: *CAD-4/PC Software* (Enraf–Nonius, 1993); cell refinement: *CAD-4/PC Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

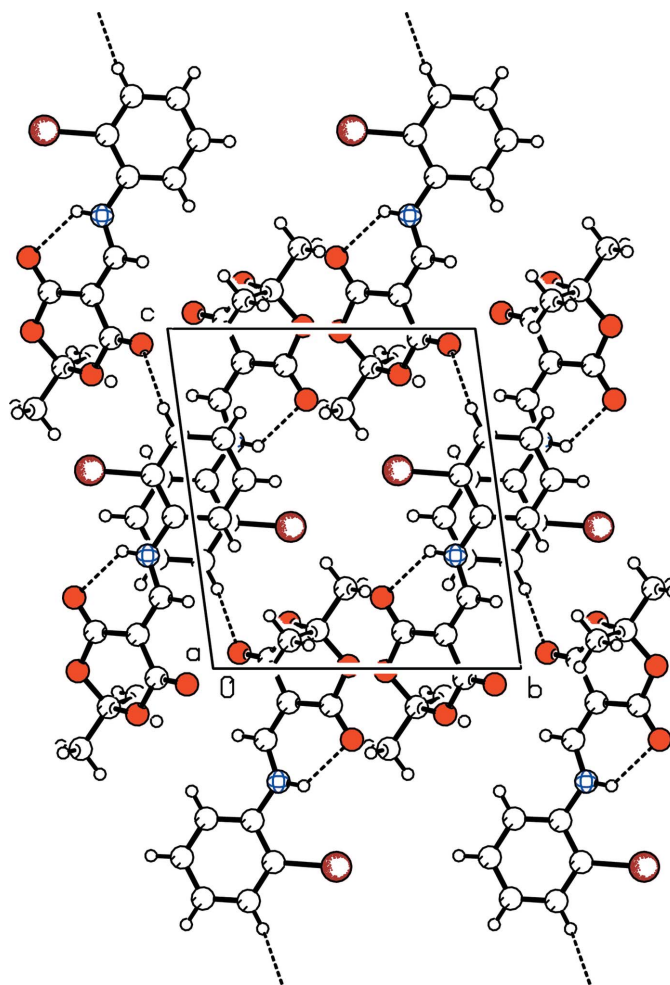


Figure 2

The molecular packing of (I), viewed along the a axis, with hydrogen bonding shown as dashed lines.

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